

Europäisches Patentamt

European Patent Office

Office européen des brevets



(11) EP 1 463 117 A2

(12)

## **EUROPEAN PATENT APPLICATION**

(43) Date of publication: 29.09.2004 Bulletin 2004/40

(51) Int Cl.7: H01L 27/00, H01L 51/20

(21) Application number: 04075861.7

(22) Date of filing: 17.03.2004

(84) Designated Contracting States:

AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
HU IE IT LI LU MC NL PL PT RO SE SI SK TR
Designated Extension States:

AL HR LT LV MK

(30) Priority: 28.03.2003 US 401658

(71) Applicant: EASTMAN KODAK COMPANY Rochester, New York 14650 (US) (72) Inventors:

- Arnold, Andrew D.,
   c/o Eastman Kodak Company
   Rochester New York 14650-2201 (FR)
- Cok, Ronald S., c/o Eastman Kodak Company Rochester New York 14650-2201 (FR)
- (74) Representative: Halle, Helen Cynthla et al Kodak Limited Patent, W92-3A, Headstone Drive Harrow, Middlesex HA1 4TY (GB)

## (54) Oled display with photosensor

(57) An OLED display including: a transparent electrode; a reflective electrode having a transparent window; a light emissive layer disposed between the trans-

parent electrode and the reflective electrode; and a photosensor located under the transparent window of the reflective electrode to sense light produced by the light emissive layer.

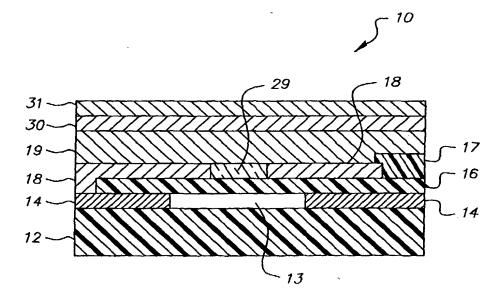


FIG. 1

## [0001] The present invention relates to solid-state

1

OLED flat-panel displays and more particularly to such displays having photosensors integrated into the display.

[0002] Solid-state organic light emitting diode (OLED) image displays are of great interest as a superior flatpanel display technology. These displays utilize current passing through thin films of organic material to generate light. The color of light emitted and the efficiency of the energy conversion from current to light are determined by the composition of the organic thin-film material. Different organic materials emit different colors of light. However, as the display is used, the organic materials in the device age and become less efficient at emitting light. It is useful, therefore, to provide a means to correct the light output of an OLED device over time. [0003] OLED devices are conventionally either topemitting or bottom-emitting. An OLED display is constructed upon a substrate by depositing an electrode upon the substrate, organic light emitting materials over the first electrode, and a second electrode above the light emitting materials. A cover is used to encapsulate and protect the device. Light is emitted by the application of a current from one electrode to another passing through the organic light emitting materials. A bottom emitting device emits light through the substrate and first electrode which must both be transparent. The second electrode may be either transparent or reflective. A topemitting device emits light through the cover and second electrode which must both be transparent. In this case, the second electrode may be either transparent or reflective.

[0004] OLED devices emit light in every direction. A portion of the light is emitted directly toward the front of the display device: through the substrate (for a bottom emitter device) or the encapsulating cover (for a top emitter device). A similar portion of the light is emitted toward the back of the display device and may be either absorbed or reflected by the electrode or other layers behind the organic layers. If the portion of light emitted toward the back is reflected, it can pass through the organic layers again and be emitted through the front of the display, thereby increasing the brightness of the display. If the portion of the light emitted toward the back is absorbed, it is absorbed and lost, thereby reducing the light emitted by one half.

[0005] Because OLED materials age, it is known to calibrate OLED display devices through the use of external sensors which measure the light output from the display device and construct a calibration table for use by the device to correct for aging. See for example US 5,371,537, issued December 6, 1994 to Bohan et al. This approach has the problem that the sensor device obscures the display during the calibration and is not capable of providing real time operation. Moreover, these approaches are not useful for correcting uniform-

ity variations among individual pixel display elements. [0006] Alternative methods utilize a light sensor integrated with the light emitting elements of the display themselves. For example, US 6,489,631, issued December 3, 2002, to Young et al. describes the integration of a photosensitive device with an electro-luminescent pixel element. The light sensing elements each comprise a gated photosensitive thin-film device such as a TFT structure having a semiconductor layer with contact regions laterally spaced on the substrate and separated by a gate controlled region. A part of the associated display element extends over the gate controlled region with an electrode of the display element serving as the gate of the photosensitive device, thereby ensuring good optical coupling between the display element and the photosensitive device. This arrangement requires the use of a transparent electrode and therefore fails to optimize emission of the light produced by the display element and limits the power that can be passed through the electrode.

[0007] There is a need therefore for an improved OLED display having integrated photosensors.

[0008] The need is met according to the present invention by providing an OLED display that includes: a transparent electrode; a reflective electrode having a transparent window; a light emissive layer disposed between the transparent electrode and the reflective electrode; and a photosensor located under the transparent window of the reflective electrode to sense light produced by the light emissive layer.

**[0009]** The advantages of this invention are an OLED display device that provides a means to measure the light output of the OLED display device while maximizing the light output.

Fig. 1 is a cross sectional schematic diagram of a light emitting element in an OLED display according to the present invention;

Fig. 2 is a cross sectional schematic diagram of a portion of a prior art top-emitting OLED display; and Fig. 3 is a schematic diagram of a prior art OLED structure.

[0010] Referring to Fig. 2, a prior art top-emitting OLED display device 10 is shown with a substrate 12, and a thin-film transistor (TFT) active matrix layer 14 comprising an array of TFTs that provides power to OLED elements. A patterned first insulating layer 16 is provided over the TFT active matrix layer 14, and an array of reflective electrodes 18 are provided over insulating layer 16 and in electrical contact with the TFT active matrix layer 14. A patterned second insulating layer 17 is provided over the array of reflective electrodes 18 such that at least a portion of the each of the reflective electrodes 18 is exposed.

[0011] Over the first electrodes and insulating layers are provided red, green, and blue-emitting organic electroluminescent (EL) elements 19. Herein, the collection

of organic EL elements may also be referred to as the organic EL layer. This layer may be made up of other layers as is known in the art, for example a hole-injection layer 32, hole-transport layer 34, emissive layer 36, electron-transport layer 38, and electron injection layer 39. Alternatively, a single continuous white emitting layer may be used with an array of color filters in the place of separate red, green and blue light emitting elements. [0012] The light-emitting pixel area is generally defined by the area of the first electrode 18 in contact with the organic EL elements. Over the organic EL layer is provided a transparent, common transparent electrode 30 that has sufficient optical transparency to allow transmission of the generated red, green, and blue light. Each first electrode in combination with its associated organic EL element and second electrode is herein referred to as an OLED element. A typical top-emitting OLED display device comprises an array of OLED elements wherein each OLED element emits red, green or blue. However, monochrome display devices are also known where the array of OLED elements emit the same color light, for example, white. An encapsulating or protective layer 31 may be provided over the transparent electrode 30.

[0013] In operation, the thin-film transistors in TFT layer 14 allow current to flow between the first electrode 18, each of which can be selectively addressed, and the common second electrode 30. Holes and electrons recombine within the organic EL elements to emit light. [0014] Referring to Fig. 1, according to one embodiment of the present invention, a photosensor 13 is located beneath the first electrode 18 and first insulating layer 16. The first insulating layer 16 is transparent and may be composed, for example, of silicon dioxide. First electrode 18 is reflective to maximize light output from the device. A transparent window 29 is provided within the first electrode 18 to allow light to pass through the opening and the transparent first insulating layer 16 to the photosensor 13. Generally, the transparent window 29 is less than half the size of the first electrode 18 and preferably smaller. Suitable electrode materials are known in the art and may include, for example, silver, aluminum, or compounds of siver and aluminum.

[0015] In operation, light is emitted by the organic materials located above the first electrode 18. Although light may not be emitted from the electroluminescent elements 19 directly above the transparent window 29, light is emitted in every direction from the areas above the first electrode 18. Some of this light will pass through the transparent window 29 to the photosensitive device 13. The light can then be measured and used by the display, for example, to calibrate or control the light output of the light emitting element using means known in the art and referred to in this disclosure. Because the transparent window 29 is much smaller than the first electrode 18, the reduction in light output from the area above the transparent window 29 is small compared to the light output achieved by the use of a reflective first

electrode 18. Because the first electrode is reflective and can be very thick, it can carry larger amounts of current than if it was constrained to be transparent. Moreover, a wider variety of materials may be used for the first electrode 18, easing the manufacturing process. [0016] The transparent window 29 can be patterned at the same time the first electrode 18 is created so that no further photolithography steps are necessary to create the opening. The window may be filled in with an insulating, transparent material, such as silicon dioxide, and preferably comprises the same material and is deposited at the same time as the second insulating layer 17. Alternatively, the organic materials 19 may be deposited directly over the window 29 with no intervening material deposited, simplifying the masking of the second insulating layer 17. In yet another alternative, a thin transparent electrode (for example, a thin metal layer made of silver or aluminum, or compounds of silver or aluminum, or a thin layer of ITO) may be provided in the transparent window 29. The thin transparent electrode may cover the entire electrode 18 or only the transparent window 29. This enhances light output and optical coupling to the photosensor 13 without significantly compromising the reflectivity and light output of the display. [0017] The size of the transparent window 29 may be adjusted to suit the sensitivity of the photosensor 13, the efficiency of the light emitting elements, and the noise requirements of the display circuitry. Alternatively, if a semitransparent opening is provided, for example with a thin conductive metal layer, the thickness (and conductivity) of the window 29 may be adjusted to optimize the light output and the photosensitive signal.

[0018] One photosensor 13 may be provided per light emitting element and the light emitting elements being deposited in a rectangular arrangement over the display. Alternatively, multiple photosensor 13 and transparent openings 29 may be provided, especially for large light emitting elements.

[0019] In a preferred embodiment, the invention is employed in a device that includes Organic Light Emitting Diodes (OLEDs) which are composed of small molecule or polymeric OLEDs as disclosed in but not limited to US 4,769,292, issued September 6, 1988 to Tang et al., and US 5,061,569, issued October 29, 1991 to Van-Slyke et al. Many combinations and variations of organic light emitting displays can be used to fabricate such a device.

[0020] The present invention can be employed in most OLED device configurations. These include very simple structures comprising a single anode and cathode to more complex devices, such as passive matrix displays comprised of orthogonal arrays of anodes and cathodes to form light emitting elements, and active-matrix displays where each light emitting element is controlled independently, for example, with thin film transistors (TFTs).

[0021] There are numerous configurations of the organic layers wherein the present invention can be suc-

55

cessfully practiced. A typical prior art structure is shown in Fig. 3 and is comprised of a substrate 101, an anode 103, a hole-injecting layer 105, a hole-transporting layer 107, a light-emitting layer 109, an electron-transporting layer 111, and a cathode 113. These layers are described in detail below. Note that the substrate may alternatively be located adjacent to the cathode, or the substrate may actually constitute the anode or cathode. The organic layers between the anode and cathode are conveniently referred to as the organic EL element. The total combined thickness of the organic layers is preferably less than 500 nm.

[0022] The anode and cathode of the OLED are connected to a voltage/current source 250 through electrical conductors 260. The OLED is operated by applying a potential between the anode and cathode such that the anode is at a more positive potential than the cathode. Holes are injected into the organic EL element from the anode and electrons are injected into the organic EL element at the anode. Enhanced device stability can sometimes be achieved when the OLED is operated in an AC mode where, for some time period in the cycle, the potential bias is reversed and no current flows. An example of an AC-driven OLED is described in US 5,552,678.

[0023] The OLED device of this invention is typically provided over a supporting substrate where either the cathode or anode can be in contact with the substrate. The electrode in contact with the substrate is conveniently referred to as the bottom electrode. Conventionally, the bottom electrode is the anode, but this invention is not limited to that configuration. The substrate can either be transmissive or opaque. In the case wherein the substrate is transmissive, a reflective or light absorbing layer is used to reflect the light through the cover or to absorb the light, thereby improving the contrast of the display. Substrates can include, but are not limited to, glass, plastic, semiconductor materials, silicon, ceramics, and circuit board materials. Of course it is necessary to provide a light-transparent top electrode.

[0024] When EL emission is viewed through anode 103, the anode should be transparent or substantially transparent to the emission of interest. Common transparent anode materials used in this invention are indium-tin oxide (ITO), indium-zinc oxide (IZO) and tin oxide, but other metal oxides can work including, but not limited to, aluminum- or indium-doped zinc oxide, magnesium-indium oxide, and nickel-tungsten oxide. In addition to these oxides, metal nitrides, such as gallium nitride, and metal selenides, such as zinc selenide, and metal sulfides, such as zinc sulfide, can be used as the anode. For applications where EL emission is viewed only through the cathode electrode, the transmissive characteristics of anode are immaterial and any conductive material can be used, transparent, opaque or reflective. Example conductors for this application include, but are not limited to, gold, iridium, molybdenum, palladium, and platinum. Typical anode materials, transmissive or

otherwise, have a work function of 4.1 eV or greater. Desired anode materials are commonly deposited by any suitable means such as evaporation, sputtering, chemical vapor deposition, or electrochemical means. Anodes can be patterned using well-known photolithographic processes. Optionally, anodes may be polished prior to application of other layers to reduce surface roughness so as to minimize shorts or enhance reflectivity.

[0025] While not always necessary, it is often useful to provide a hole-injecting layer 105 between anode 103 and hole-transporting layer 107. The hole-injecting material can serve to improve the film formation property of subsequent organic layers and to facilitate injection of holes into the hole-transporting layer. Suitable materials for use in the hole-injecting layer include, but are not limited to, porphyrinic compounds as described in US 4,720,432, plasma-deposited fluorocarbon polymers as described in US 6.208.075, and some aromatic amines, for example, m-MTDATA (4,4',4"-tris[(3-methylphenyl)phenylamino]triphenylamine). Alternative hole-injecting materials reportedly useful in organic EL devices are described in EP 0 891 121 1A and EP 1029 909 A1.

[0026] The hole-transporting layer 107 contains at least one hole-transporting compound such as an aromatic tertiary amine, where the latter is understood to be a compound containing at least one trivalent nitrogen atom that is bonded only to carbon atoms, at least one of which is a member of an aromatic ring. In one form the aromatic tertiary amine can be an arylamine, such as a monoarylamine, diarylamine, triarylamine, or a polymeric arylamine. Exemplary monomeric triarylamines are illustrated by Klupfel et al. US 3,180,730. Other suitable triarylamines substituted with one or more vinyl radicals and/or comprising at least one active hydrogen containing group are disclosed by Brantley et al. in US 3,567,450 and 3,658,520.

[0027] A more preferred class of aromatic tertiary amines are those which include at least two aromatic tertiary amine moieties as described in US 4,720,432 and 5,061,569. The hole-transporting layer can be formed of a single or a mixture of aromatic tertiary amine compounds. Illustrative of useful aromatic tertiary 45 amines are the following:

- 1,1-Bis(4-di-p-tolylaminophenyl)cyclohexane
- 1,1-Bis(4-di-p-tolylaminophenyl)-4-phenylcy-clohexane
- 4,4'-Bis(diphenylamino)quadriphenyl Bis(4-dimethylamino-2-methylphenyl)-phenylmethane

N,N,N-Tri(p-tolyl)amine

4-(di-p-tolylamino)-4'-[4(di-p-tolylamino)-styryl]stil-

N,N,N',N'-Tetra-p-tolyl-4-4'-diaminobiphenyl N,N,N',N'-Tetraphenyl-4,4'-diaminobiphenyl N,N,N',N'-tetra-1-naphthyl-4,4'-diaminobiphenyl N,N,N',N'-tetra-2-naphthyl-4,4' -diaminobiphenyl N-Phenylcarbazole

4,4'-Bis[N-(1-naphthyl)-N-phenylamino]biphenyl 4,4'-Bis[N-(1-naphthyl)-N-(2-naphthyl)amino]biphenyl

4,4"-Bis[N-(1-naphthyl)-N-phenylamino]p-terphenyl

4,4'-Bis[N-(2-naphthyl)-N-phenylamino]biphenyl 4,4'-Bis[N-(3-acenaphthenyl)-N-phenylamino]biphenyl

1,5-Bis[N-(1-naphthyl)-N-phenylamino]naphthalene

4,4'-Bis[N-(9-anthryl)-N-phenylamino]biphenyl 4,4"-Bis[N-(1-anthryl)-N-phenylamino]-p-terphenyl 4,4'-Bis[N-(2-phenanthryl)-N-phenylamino]biphenyl

4,4'-Bis[N-(8-fluoranthenyl)-N-phenylamino]biphenyl

4,4'-Bis[N-(2-pyrenyl)-N-phenylamino]biphenyl 4,4'-Bis[N-(2-naphthacenyl)-N-phenylamino]biphenyl

4,4'-Bis[N-(2-perylenyl)-N-phenylamino]biphenyl

4,4'-Bis[N-(1-coronenyl)-N-phenylamino]biphenyl

2,6-Bis(di-p-tolylamino)naphthalene

2,6-Bis[di-(1-naphthyl)amino]naphthalene

2,6-Bis[N-(1-naphthyl)-N-(2-naphthyl)amino]naphthalene

N,N,N',N'-Tetra(2-naphthyl)-4,4"-diamino-p-terphenyl

4,4'-Bis {N-phenyl-N-[4-( 1-naphthyl)-phenyl] amino}biphenyl

4,4'-Bis[N-phenyl-N-(2-pyrenyl)amino]biphenyl

2,6-Bis[N,N-di(2-naphthyl)amine]fluorene

1,5-Bis[N-(1-naphthyl)-N-phenylamino]naphthalene

4,4',4"-tris[(3-methylphenyl)phenylamino]triphenylamine

[0028] Another class of useful hole-transporting materials includes polycyclic aromatic compounds as described in EP 1 009 041. Tertiary aromatic amines with more than two amine groups may be used including oligomeric materials. In addition, polymeric hole-transporting materials can be used such as poly(N-vinylcarbazole) (PVK), polythiophenes, polypyrrole, polyaniline, and copolymers such as poly(3,4-ethylenedioxythiophene) / poly(4-styrenesulfonate) also called PEDOT/ PSS.

[0029] As more fully described in US 4,769,292 and 5,935,721, the light-emitting layer (LEL) 109 of the organic EL element includes a luminescent or fluorescent material where electroluminescence is produced as a result of electron-hole pair recombination in this region. The light-emitting layer can be comprised of a single material, but more commonly consists of a host material doped with a guest compound or compounds where light emission comes primarily from the dopant and can be of any color. The host materials in the light-emitting

layer can be an electron-transporting material, as defined below, a hole-transporting material, as defined above, or another material or combination of materials that support hole-electron recombination. The dopant is usually chosen from highly fluorescent dyes, but phosphorescent compounds, e.g., transition metal complexes as described in WO 98/55561, WO 00/18851, WO 00/57676, and WO 00/70655 are also useful. Dopants are typically coated as 0.01 to 10 % by weight into the host material. Polymeric materials such as polyfluorenes and polyvinylarylenes (e.g., poly(p-phenylenevinylene), PPV) can also be used as the host material. In this case, small molecule dopants can be molecularly

dispersed into the polymeric host, or the dopant could
 be added by copolymerizing a minor constituent into the host polymer.

[0030] An important relationship for choosing a dye as a dopant is a comparison of the bandgap potential which is defined as the energy difference between the highest occupied molecular orbital and the lowest unoccupied molecular orbital of the molecule. For efficient energy transfer from the host to the dopant molecule, a necessary condition is that the band gap of the dopant is smaller than that of the host material. For phosphorescent emitters it is also important that the host triplet energy level of the host be high enough to enable energy transfer from host to dopant.

[0031] Host and emitting molecules known to be of use include, but are not limited to, those disclosed in US 4,768,292; 5,141,671; 5,150,006; 5,151,629; 5,405,709; 5,484,922; 5,593,788; 5,645,948; 5,683,823; 5,755,999; 5,928,802; 5,935,720; 5,935,721; and 6,020,078.

[0032] Metal complexes of 8-hydroxyquinoline (oxine) and similar derivatives constitute one class of useful host compounds capable of supporting electroluminescence. Illustrative of useful chelated oxinoid compounds are the following:

CO-1: Aluminum trisoxine [alias, tris(8-quinolinolato)aluminum(III)]

CO-2: Magnesium bisoxine [alias, bis(8-quino-linolato)magnesium(II)]

CO-3: Bis[benzo{f}-8-quinolinolato]zinc (II)

CO-6: Aluminum tris(5-methyloxine) [alias, tris (5-methyl-8-quinolinolato) aluminum(III)]

CO-7: Lithium oxine [alias, (8-quinolinolato)lithium (I)]

CO-8: Gallium oxine [alias, tris(8-quinolinolato)gal-lium(III)]

 CO-9: Zirconium oxine [alias, tetra(8-quinolinolato) zirconium(IV)]

[0033] Other classes of useful host materials include,

40

but are not limited to: derivatives of anthracene, such as 9,10-di-(2-naphthyl)anthracene and derivatives thereof as described in US 5,935,721, distyrylarylene derivatives as described in US 5,121,029, and benzazole derivatives, for example, 2, 2', 2"-(1,3,5-phenylene)tris [1-phenyl-1H-benzimidazole]. Carbazole derivatives are particularly useful hosts for phosphorescent emitters

[0034] Useful fluorescent dopants include, but are not limited to, derivatives of anthracene, tetracene, xanthene, perylene, rubrene, coumarin, rhodamine, and quinacridone, dicyanomethylenepyran compounds, thiopyran compounds, polymethine compounds, pyrilium and thiapyrilium compounds, fluorene derivatives, periflanthene derivatives, indenoperylene derivatives, bis (azinyl)amine boron compounds, bis(azinyl)methane compounds, and carbostyryl compounds.

[0035] Preferred thin film-forming materials for use in forming the electron-transporting layer 111 of the organic EL elements of this invention are metal chelated oxinoid compounds, including chelates of oxine itself (also commonly referred to as 8-quinolinol or 8-hydroxyquinoline). Such compounds help to inject and transport electrons, exhibit high levels of performance, and are readily fabricated in the form of thin films. Exemplary oxinoid compounds were listed previously.

[0036] Other electron-transporting materials include various butadiene derivatives as disclosed in US 4,356,429 and various heterocyclic optical brighteners as described in US 4,539,507. Benzazoles and triazines are also useful electron-transporting materials.

[0037] When light emission is viewed solely through the anode, the cathode 113 used in this invention can be comprised of nearly any conductive material. Desirable materials have good film-forming properties to ensure good contact with the underlying organic layer, promote electron injection at low voltage, and have good stability. Useful cathode materials often contain a low work function metal (< 4.0 eV) or metal alloy. One preferred cathode material is comprised of a Mg:Ag alloy wherein the percentage of silver is in the range of 1 to 20 %, as described in US 4,885,221. Another suitable class of cathode materials includes bilayers comprising a thin electron-injection layer (EIL) in contact with the organic layer (e.g., ETL) which is capped with a thicker layer of a conductive metal. Here, the EIL preferably includes a low work function metal or metal salt, and if so, the thicker capping layer does not need to have a low work function. One such cathode is comprised of a thin layer of LiF followed by a thicker layer of A1 as described in US 5,677,572. Other useful cathode material sets include, but are not limited to, those disclosed in US 5,059,861, 5,059,862, and 6,140,763.

[0038] When light emission is viewed through the cathode, the cathode must be transparent or nearly transparent. For such applications, metals must be thin or one must use transparent conductive oxides, or a combination of these materials. Optically transparent

cathodes have been described in more detail in US 4,885,211, US 5,247,190, JP 3,234,963, US 5,703,436, US 5,608,287, US 5,837,391, US 5,677,572, US 5,776,622, US 5,776,623, US 5,714,838, US 5,969,474, US 5,739,545, US 5,981,306, US 6,137,223, US 6,140,763, US 6,172,459, EP 1 076 368, US 6,278,236, and US 6,284,393. Cathode materials are typically deposited by evaporation, sputtering, or chemical vapor deposition. When needed, patterning can be achieved through many well known methods including, but not limited to, through-mask deposition, integral shadow masking, for example, as described in US 5,276,380 and EP 0 732 868, laser ablation, and selective chemical vapor deposition.

[0039] In some instances, layers 109 and 111 can optionally be collapsed into a single layer that serves the function of supporting both light emission and electron transportation. It also known in the art that emitting dopants may be added to the hole-transporting layer, which may serve as a host. Multiple dopants may be added to one or more layers in order to create a white-emitting OLED, for example, by combining blue- and yellow-emitting materials, cyan- and red-emitting materials, or red-, green-, and blue-emitting materials. White-emitting devices are described, for example, in EP 1 187 235, US 20020025419, EP 1 182 244, US 5,683,823, US 5,503,910, US 5,405,709, and US 5,283,182.

[0040] Additional layers such as electron or hole-blocking layers as taught in the art may be employed in devices of this invention. Hole-blocking layers are commonly used to improve efficiency of phosphorescent emitter devices, for example, as in US 20020015859.
[0041] This invention may be used in so-called stacked device architecture, for example, as taught in

US 5,703,436 and US 6,337,492.

[0042] The organic materials mentioned above are suitably deposited through a vapor-phase method such as sublimation, but can be deposited from a fluid, for example, from a solvent with an optional binder to improve film formation. If the material is a polymer, solvent deposition is useful but other methods can be used, such as sputtering or thermal transfer from a donor sheet. The material to be deposited by sublimation can be vaporized from a sublimator "boat" often comprised of a tantalum material, e.g., as described in US 6,237,529, or can be first coated onto a donor sheet and then sublimed in closer proximity to the substrate. Layers with a mixture of materials can utilize separate sublimator boats or the materials can be pre-mixed and coated from a single boat or donor sheet. Patterned deposition can be achieved using shadow masks, integral shadow masks (US 5,294,870), spatially-defined thermai dye transfer from a donor sheet (US 5,688,551, 5,851,709 and 6,066,357) and inkjet method (US 6,066,357).

[0043] Most OLED devices are sensitive to moisture or oxygen, or both, so they are commonly sealed in an inert atmosphere such as nitrogen or argon, along with 10

15

25

30

a desiccant such as alumina, bauxite, calcium sulfate, clays, silica gel, zeolites, alkaline metal oxides, alkaline earth metal oxides, sulfates, or metal halides and perchlorates. Methods for encapsulation and desiccation include, but are not limited to, those described in US 6,226,890. In addition, barrier layers such as SiOx, Teflon, and alternating inorganic/polymeric layers are known in the art for encapsulation.

[0044] OLED devices of this invention can employ various well-known optical effects in order to enhance its properties if desired. This includes optimizing layer thicknesses to yield maximum light transmission, providing dielectric mirror structures, replacing reflective electrodes with light-absorbing electrodes, providing anti glare or anti-reflection coatings over the display, providing a polarizing medium over the display, or providing colored, neutral density, or color conversion filters over the display. Filters, polarizers, and anti-glare or anti-reflection coatings may be specifically provided over the cover or an electrode protection layer beneath the cover.

sensor is a thin film device.

- The display claimed in claim 1, wherein the window is located in the center of the reflective electrode.
- The display claimed in claim 1, wherein the window is located at an edge of the reflective electrode.
- The display claimed in claim 1, wherein the reflective electrode includes a plurality of windows.
- The display claimed in claim 11, further comprising a plurality of photosensors corresponding to the plurality of windows.

## Claims

- 1. An OLED display, comprising:
  - a) a transparent electrode;
  - b) a reflective electrode having a transparent window:
  - c) a light emissive layer disposed between the transparent electrode and the reflective electrode; and
  - d) a photosensor located under the transparent window of the reflective electrode to sense light produced by the light emissive layer.
- The display claimed in claim 1, wherein the transparent window is an opening in the reflective electrode.
- The display claimed in claim 2, wherein the opening is filled with a transparent insulator.
- The display claimed in claim 3, wherein the transparent insulator is SiO<sub>2</sub>.
- The display claimed in claim 2, wherein the opening is filled with transparent conductor.
- The display claimed in claim 5, wherein the transparent conductor is ITO.
- The display claimed in claim 1, wherein the transparent window is a thin region in the reflective electrode.
- 8. The display claimed in claim 1, wherein the photo-

50

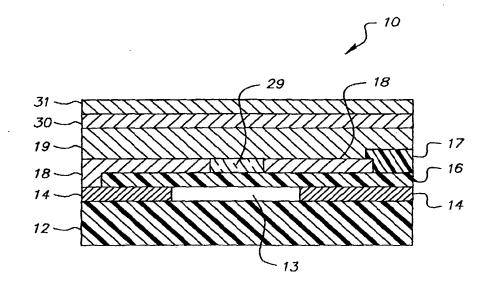


FIG. 1

